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PERIODIC TABLE OF THE ELEMENTS & THEIR ATOMS

KEY

ATOMIC NUMBER

- Naturally Radioactive
- Synthetically Radioactive

OXIDATION STATES

SYMBOL (Note 1)

- Solid
- Liquid (Note 3)
- Gas

NAME

1 1.00
H
Hydrogen
1s¹

1
-252.7
-259.2
2.2

1s¹

ATOMIC WEIGHT (Note 2)

BOILING POINT °C

MELTING POINT °C

ELECTRONEGATIVITY (PAULING'S)

GROUND STATE ELECTRON CONFIGURATION

METAL

NON-METAL

METALLOID

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NOTES

- Element symbols, names and spelling as recommended by IUPAC® (International Union of Pure and Applied Chemistry).
- Atomic weight: Mean relative masses of naturally occurring isotopic mixtures based on ¹²C as recommended by IUPAC® 2013. A number in parenthesis indicates the uncertainty in the last digit of the atomic weight. Elements with no stable nuclides have the value for the longest lived isotope given in brackets e.g. [209]. Those elements with a range in their atomic weights, due to differing isotopic abundances in different parts of the world, have been given an approximate atomic weight.
- Liquid at 30°C